

Novel approaches in human PK study design (e.g. stable isotopes technique) to overcome the challenges in the conduct of dedicated BA/BE studies-Case Studies

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Disclaimer & Acknowledgements



- Disclaimer:
 - I am a statistician
 - Last biology or chemistry class was in high school
- Acknowledgements
 - Manish Gupta
 - Alan Parr
 - Frank Hoke

Agenda



- Motivation
- Describe the Stable Isotope Label approach (SIL)
- Pilot Study
- Potential Benefits
- Limitations
- Summary

Motivation



- Regulatory requirements for Quality by Design (QbD) of drug products
- Desire to link *in vitro* data (e.g. dissolution data) with *in vivo* data (e.g. human BA/BE studies)
- Traditional approaches to human BA/BE studies may not be feasible in many cases due required number of subjects

Re-introduce a Novel approach

Use of Stable Isotopes (SIL) in Pharmacokinetic Studies

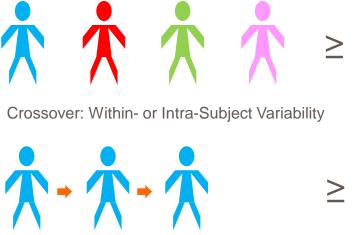
- Use of stable isotopes (has been in existence for many years.
- Provides the ability to measure plasma concentrations of an enriched and nonenriched drug substance from the same plasma sample (i.e., subject).
- The idea:
 - Dose two formulations at the same time in the same subject
 - Result in very small variability [Heck et al, 1979]

Parallel Grp: Between- or Inter-Subject Variability

SIL: Within- or Intra-Subject/period Variability

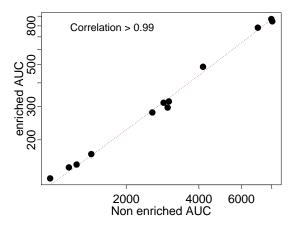
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Stable Isotope Approach

- Use the label drug (or SIL) as an internal control [Parr et al. 2012]
 - each subject at each dosing period will receive a small dose of the compound in question containing enriched isotope
- 2 PK parameters for each subject/dosing period (non-enriched and SIL)
 - Should be highly correlated ($\rho > 0.95$)
- Analysis "adjust" for the SIL resulting in the variability of the statistical test being reduced and subsequently reducing the required sample size



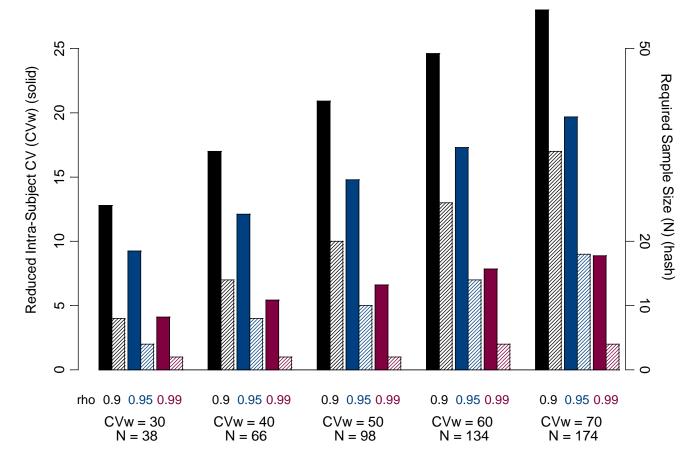


SIL – reduction in variability/sample size

gsk

Source Parr et al. 2012





For each combination of CV_w and rho, 1000 trials of 16 subjects were simulated and average residual mean square error (RMSE) was calculated. The PK parameters for both test and reference products and for the stable isotope was assumed to have a mean of 500, 500 and 50, respectively. The average RMSE presented as the reduced CV_w and the associated sample size required to provide 90% power to demonstrate bioequivalence are provided.





- Methods is only appropriate under the assumptions that
 - Formulations that are qualitatively and quantitatively the same and
 - Subject/period are qualitatively and quantitatively the same
- What does this mean? May not be valid for
 - Formulations with different components
 - Drug interaction studies
 - Food effect studies
 - etc

Objectives/Design



- Objectives
 - Known case of equivalence
 - Known case of in-equivalence
 - Known case of food effect
- 4 period crossover
 - Regimen A and B were the same oral formulation and dose
 - Regimen C same as oral formulation as A but 25% higher dose
 - Regimen D same as oral formulation as A but given with food
 - SIL was an aqueous solution and 10% dose of A
- Planned sample size was 12 Normal Healthy Volunteers
 - Assume correlation of 0.95
- Analyzed both as a crossover and parallel group (using period 1 only, n = 3).

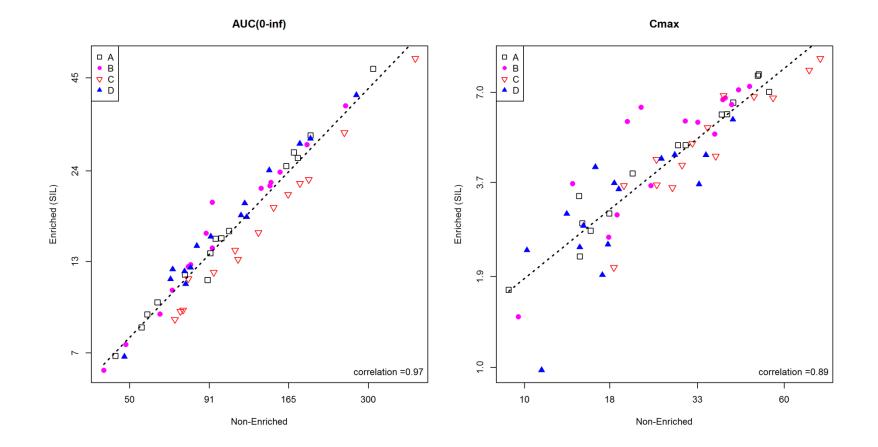
Results



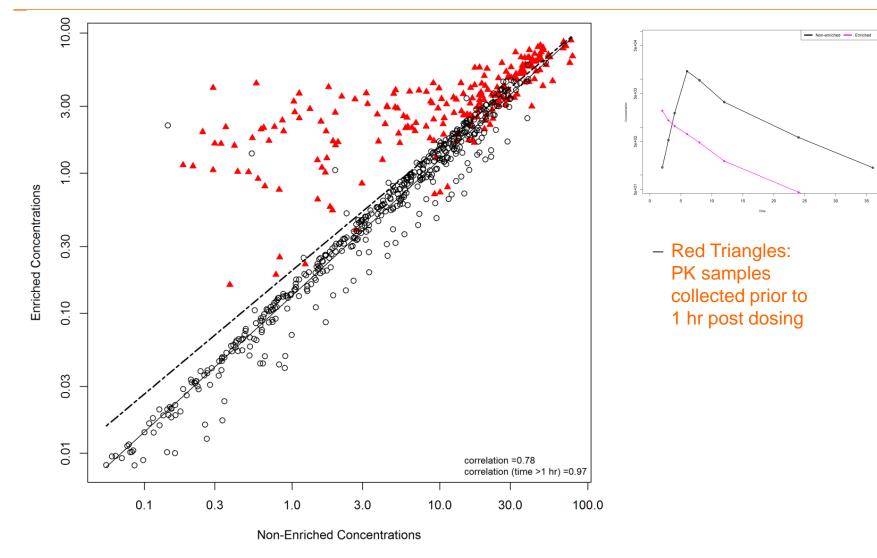
- AUC correlation was as expected, but Cmax was lower (0.85-0.89).
- Cross-over
 - Large reduction in variability
 - Correct conclusions regarding BE, B:A was equivalent, C:A was in-equivalent.
 - Incorrect conclusions regarding food effect (D:A), a lack of effect was demonstrated for Cmax
- Parallel group
 - Large reduction in variability
 - Point estimates of ratios varied from expected values and thus resulted in failure to draw correct inference in some cases.
 - Similar results when data from periods 2, 3 & 4 were analyzed

Lower than expected correlation in Cmax











Incorrect inference in Food Effect

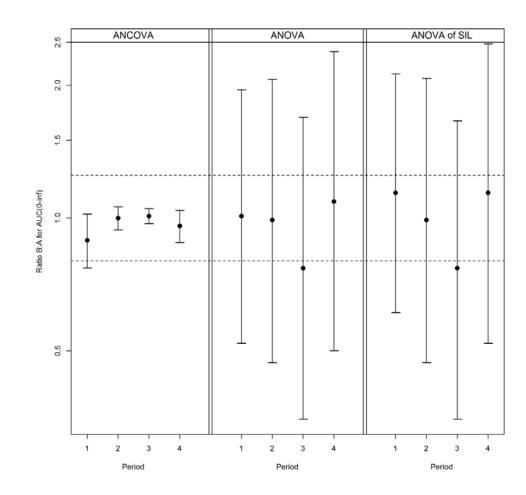


	Geometric	Geometric		90%	
Parameter	Lsmean	LSmean		Confidence	
	Test Tmt.	Ref. Tmt.	Ratio	Interval	CVw%
Cmax(units)	25892.1	25253.2	1.03	(0.89,1.18)	21.83
Cmax(units)	19484.4	25907.3	0.75	(0.61,0.92)	33.55
Cmax(units)	3087.9	4259.8	0.72	(0.63,0.84)	23.90
			$ \bigcirc $		

- -Food decreased the absorption of both non-enriched and SIL formulations
- -Adjusting for the SIL in the model, masks the food effect
- -Ratio of 0.75 / 0.72 = 1.04



Inconsistent results in parallel group





Advantages

- a. Enriched isotope acts as reference within each subject/period reducing variability
- b. Decreases number of subjects required to achieve desired power
- c. Reduces number of dosing periods (when compared to the current replicate design approach for highly variable drugs)
- d. Reduces overall cost of PK studies
- e. Does not require the manufacture of large amounts of enriched drug substance
- f. Does not require the manufacture of formulations containing the enriched isotope
- g. Reduces number of PK samples that need to be analyzed
- h. Doesn't use radioactivity

Stable Isotope Limitations



Limitations

- a. Need to confirm that there is no isotope effect (usually not seen if we don't use deuterium)
 - Satisfy the "qualitatively and quantitatively the same" assumption
- b. Need to have enriched isotope compound synthesized to do these studies (estimated cost is 10K pounds & 12 weeks)
- c. May need to synthesize two different enriched isotopes if one is needed as an internal standard for analytical analysis purposes (not an issue since multiple compounds can be made)
- d. Time and cost to synthesize these compounds so that the labelled site is metabolically stable (this can be built into the project plans)

When would you use this approach?



- For compounds that exhibit high PK variability
- For compounds where subject/ patient recruitment is difficult
- Comparison of products manufactured at different sites
- Formulation/ Process screening studies
- Animal studies where small number of animals are used
- Many others

Note: Stable isotope approach adds value even if compound exhibits acceptable bioavailability

References



Heck HA, Buttrill SE, Flynn NW, Dyer RL, Anbar M, Cairns T, Dighe S, Cabana BE. (1979). Bioavailability of imipramine tablets relative to a stable isotope-labelled internal standard: increasing the power of bioavailability tests. J Pharmacokinet Biopharm. 1979;7(3):233–48.

Parr A, Gupta M, Montague TH and Hoke F (2012). Re-introduction of a Novel Approach to the Use of Stable Isotopes in Pharmacokinetic Studies. The AAPS Journal (2012), DOI: 10.1208/s12248-012-9371-4